

1 **MACROSCOPIC BEHAVIOR AND MICROSTRUCTURAL ANALYSIS OF RECYCLED**  
2 **AGGREGATE MORTAR BARS EXPOSED TO EXTERNAL SULFATE ATTACK**

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15  
16 **Keywords:** recycled concrete aggregate, sulfate attack, salt cristalization, expansion, thermal  
17 **analyses**

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20  
21 **ABSTRACT**

22  
23 The performance of cementitious materials made with Recycled Concrete Aggregates (RCA)  
24 and exposed to sulfate-laden environments is not yet fully understood. There are few studies  
25 on this topic, and for a full description of the mechanisms of external sulfate attack (ESA),  
26 several influencing parameters should be considered. The replacement of natural aggregates  
27 by RCA modifies the physical and chemical properties of the new material and introduces  
28 additional variables into the already complex mechanism. The high porosity of RCA contributes  
29 to accelerating the sulfate penetration from the surface, while the attached mortar increases  
30 the mineral supply that can react with the incoming sulfate.

31 This paper describes a study on the performance of mortars made with RCA exposed to  
32 different ESA conditions. Design and exposure parameters include three cement, two  
33 specimen sizes, and two sulfate media (soil and solution). Visual inspection and  
34 measurements of expansion and weight change were used to determine the performance of  
35 the mixes. From a microstructural point of view, thermogravimetric analyses were carried out  
36 on samples obtained from the mortar bars after exposure. These results are then used to infer  
37 on the thermodynamic and kinetic aspects of ESA progression in recycled mortar. Finally,  
38 SEM/EDS analyses were also performed to further describe sulfate penetration and its  
39 relationship with cracking. The results show a limited influence of fine recycled aggregate on  
40 the mortar performance against ESA and no proportionality to its content.

## 1. INTRODUCTION

The production of concrete, one of the most used materials in the construction industry, causes significant environmental impacts. The concrete industry consumes a significant amount of non-renewable resources. Particularly in the case of aggregates, the depletion of resources through the exploitation of natural quarries is increasingly banned due to the impact on the landscape and the reduction of productive land. Moreover, the volume of concrete waste in landfills continues to increase. Thus, the use of concrete waste to produce aggregates for the production of new concrete is an environmentally friendly and viable practice, as it enables reductions in the mining of natural aggregate and landfilling of construction and demolition waste [1–6].

The use of recycled concrete aggregate (RCA) in the production of new concrete has been extensively studied for decades. Several reports have shown that the use of coarse RCA as a partial replacement of coarse natural aggregate has a limited effect on the mechanical properties of concrete [7–11]. It is known that the mortar attached to RCA particles increases the volume of the interfacial transition zone and the overall porosity of the new concrete [8,12–14]. These effects may affect the durability of recycled aggregate concrete, as transport properties are increased. Nevertheless, there is not a completed knowledge about the RCA effect in each type of exposition type, and some results in the literature are not consistent yet [15–20].

Regarding the specific case of external sulfate attack (ESA), there is limited literature that evaluates mixes with RCA [21–28], and these results are not entirely consistent. Some studies indicate a poorer performance of cement-based mixes with RCA [5,21,22,29,30], while others suggest a similar or even better performance of mixes with RCA [24,27,31,32].

It appears that other parameters of each different experimental campaign (materials, mixes proportioning, types of exposure, etc.) determine the effect of RCA on the ESA phenomenon. The study of ESA process involves many parameters, and the use of RCA introduces new variables that further complicate the assessment in a single experimental campaign. The use of RCA can negatively affect the performance of concrete or mortar exposed to sulfate-laden media for two main reasons: first, it increases concrete porosity, which may translate into an increase in the sulfate transport rate; second, attached mortar in RCA implies an additional supply of hydration products of Portland cement (including aluminates), which increases the reactivity with sulfates. In this sense, the subject deserves experimental studies to assess these potential effects. Finally, the higher porosity of RCA bearing mixes could play a positive role since it provides more free space for the accommodation of ESA products (buffer capacity) [33].

In addition, there are several proposed methods in the literature for the evaluation of ESA on concrete and mortar specimens. Macroscopic physical changes (e.g., length and weight variations, cracking) are the most widely used evaluation methods due to their practical implementation. However, these refer to volumetric properties, and as such they are unable to explain in depth the kinetics of the ESA process [34]. As ESA starts on the surface and progresses in depth with time, microstructural analysis and profiling are necessary to establish a correlation between macroscopic changes (length, weight, and cracking) and microscopic changes due to ESA [35–37]. Such information for RCA concrete and mortar is particularly lacking in the literature. Mineralogical changes produced by ESA can be assessed by techniques such as SEM/BSE, XRD, or TGA.

101  
 102 The type of exposure of samples can also influence the evaluation of the behavior of recycled  
 103 mortars against ESA. The saturation state of samples during testing could determine the  
 104 transport mechanism of the ions into the matrix, and the porosity involved in it. In a saturated  
 105 state, the transport mechanism is diffusion, and the entire porosity is affected. In a non-  
 106 saturated state, capillary absorption also has a participation rate, but only a part of the total  
 107 porosity takes part [38–40]. In addition, in a non-saturated state, ESA is likely to be combined  
 108 with physical sulfate attack (PSA), which is a particular form of salt crystallization [41–43]. The  
 109 role of RCA may also change regarding the predominant attack mechanism, since it modifies  
 110 the porosity system of the evaluated material and other related properties, such as mechanical  
 111 strength and stiffness. The size and slenderness of specimens also have an effect on ESA  
 112 progression [35,44], since the ESA and PSA have a superficial characteristic.

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 114 The present study provides results of experimental procedures with the aim of knowing, on the  
 115 one hand, the influence of RCA on the ESA process and, on the other hand, assessing possible  
 116 methods to evaluate the potential effect of RCA mortar under ESA. Mortar mixes with different  
 117 contents of fine recycled aggregate (FRA) and cement types were prepared. Two specimen  
 118 sizes were cast with each mortar and exposed to two conditions: immersion in sulfate solution,  
 119 and burial in sulfate-rich soil. Common evaluation methods, such as length and weight  
 120 changes and visual inspection, were used to determine the ESA-related damage on samples.  
 121 In addition, microstructural analyses were performed to determine the effects of the variables  
 122 considered on the test results and confirm the macroscopic changes observed.

123  
 124 **2. METHODOLOGY**

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 126 **2.1. Materials and specimens**

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 128 Three types of Portland cement were used in this study: a low C<sub>3</sub>A Ordinary Portland cement  
 129 (O) (equivalent to CEM I 42.5 N, EN 197-1); a low-C<sub>3</sub>A limestone-blended Portland cement (L)  
 130 (equivalent to CEM II/A-L 42.5 N, EN 197-1); and a high-C<sub>3</sub>A limestone-blended Portland  
 131 cement (H) (equivalent to CEM II/A-L 52.5 N, EN 197-1). Table 1 summarizes the chemical  
 132 and mineralogical compositions of the cements as determined by XRF and Rietveld analyses  
 133 [45], respectively.

134  
 135 Table 1 - Chemical and mineralogical compositions of cements.

Cem.	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	SO <sub>3</sub>	C <sub>3</sub> S	C <sub>2</sub> S	C <sub>3</sub> A	C <sub>4</sub> AF	C	G	LF
O	64.2	18.9	3.7	2.9	2.3	18	56	4.9	8.8	87.7	4.9	4.2
L	65.8	18.8	3.1	3.0	1.9	13	41	3.2	7.8	65.0	3.9	22.0
H	65.9	19.2	3.6	0.1	2.6	18	50	9.4	0.5	77.9	3.7	17.2

C: Total clinker; G: Gypsum; LF: Limestone Filler

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 137 The fine recycled aggregate (FRA) was obtained from crushing waste concrete with a  
 138 compressive strength of 25 MPa (i.e., structural concrete). Natural river siliceous sand (NS)  
 139 and manufactured sand (crushed granite) (CGS) were used as fine aggregates. Table 2  
 140 presents the properties of the fine aggregates used.

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 143

144 Table 2 – Properties of fine aggregates.

Sand	Water Absorption, (%)	Fineness modulus	Density, (gr/cm <sup>3</sup> )
NS	0.4	1.06	2.65
CGS	0.6	3.59	2.69
FRA	5.8	3.55	2.30

145  
 146 Table 3 presents the mix proportions for each mortar. FRA mixes contained 30 or 50 % vol. of  
 147 FRA with respect to the total volume of fine aggregate, with the remainder being NS. The  
 148 control mixes were made with 30 % vol. CGS and NS for the rest. Each mix is labelled with a  
 149 letter indicating the constituting cement and a number indicating the FRA relative content.

150  
 151 The mortars were designed with a water-to-cement ratio of 0.45, a value usually recommended  
 152 for producing sufficiently durable concrete. For the FRA mortars, mixing water content was  
 153 adjusted by adding an amount equal to 50% of the FRA's water absorption capacity. A dose  
 154 of superplasticizer was added in all mixes to obtain good flowability, using the same admixture  
 155 in 1% of the weight of cement. This resulted in minor variations in the flowability of mixes, but  
 156 in all cases, the workability was sufficient for proper casting and compaction in the molds. A  
 157 standard high energy compaction procedure with a jolting table was applied to all samples,  
 158 and the variations in consistency did not result in variations of the degree of compaction.

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 160 Table 3 - Proportions of mortars (g).

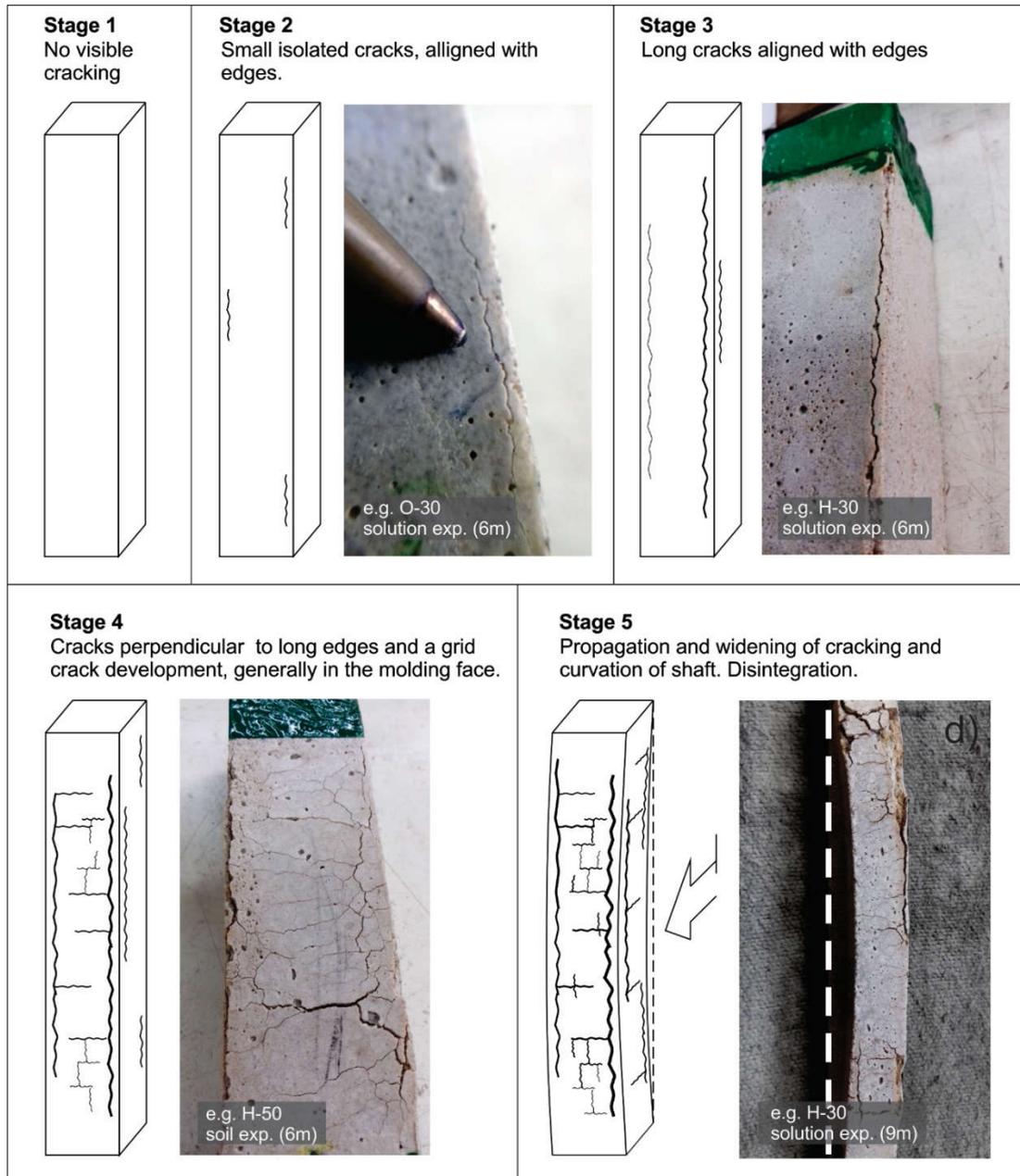
Mortar ID	Cement	Water	NS	CGS	FRA
O-0	1000	450	1806	774	-
O-30	1000	477	1806	-	669
O-50	1000	495	1280	-	1135
L-0	1000	450	1806	774	-
L-30	1000	477	1806	-	669
L-50	1000	495	1280	-	1135
H-0	1000	450	1806	774	-
H-30	1000	477	1806	-	669
H-50	1000	495	1280	-	1135

161  
 162 For each mortar, six prisms of 25 x 25 x 285 mm<sup>3</sup> (long bars) and six prisms of 40 x 40 x 160  
 163 mm<sup>3</sup> (short bars) were made for length and weight measurements. Additionally, three prisms  
 164 of 40 x 40 x 160 mm<sup>3</sup> were cast to determine flexural and compressive strengths. All specimens  
 165 were cured in lime-saturated water at a temperature of 23 ± 2 °C for 28 days. The specimens  
 166 used for evaluating expansion and weight variation were sealed with epoxy paint at their ends  
 167 (the surface around the pins) to prevent attack near the pins. After the paint had hardened,  
 168 they were once again saturated in lime water for 72 hours before the initial measurements and  
 169 exposure to sulfates. The flexural and compressive strength were evaluated following the  
 170 Argentinean Standard IRAM 1622 [46] (similar to ASTM C 438 and ASTM C 109). The water  
 171 absorption and density (with 3 samples per mix) were evaluated using the unaffected  
 172 remaining pieces after the mechanical tests.

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## 2.2. Exposure conditions and measurements

For each type of mortar and specimen size, the six bars (three per group) were exposed to two different conditions, both corresponding to very aggressive environments as described in ACI 318-14 [47]. The first group was buried in 2 g of  $\text{Na}_2\text{SO}_4$ /100 g of soil, with a mortar-to-soil volume ratio of 1:3. The soil vats were stored at ambient temperature (varying between 15 and 25 °C), and the internal relative humidity was always above 80%. The second group was immersed in 50 g/L  $\text{Na}_2\text{SO}_4$  solution that was kept at  $20 \pm 2$  °C and pH at 6-8, with periodic corrections with diluted sulfuric acid added after each measurement as necessary.



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Figure 1 – Stages of ESA progression.

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All mortar bars were evaluated by means of length and weight variations after exposure periods of 0, 7, 14, 28 days, and every month up to 1 year of exposure. For the purpose of visual inspection, five stages of ESA progression were stated, as indicated in Figure 1. With each measurement, each specimen was qualified visually and classified in the corresponding stage (it was considered that a group reached each stage when 2 of the 3 specimens in the group reached that stage).

**2.3. Samples extraction and instrumental analysis**

For the thermogravimetric analyses, a representative specimen was selected for each series, and two samples were collected manually as shown in Figure 2. For each mortar specimen, both bulk and surface samples were mechanically grounded and immediately evaluated using a NIEZTCH Jupiter device (from 20 to 1050 °C at 10 °C/min in an N<sub>2</sub> atmosphere). TG and DTG curves were analyzed to quantify compositions. Furthermore, polished section samples were prepared from the middle-part of the same representative mortar bars, and analyzed with scanning electron microscopy with energy dispersive X-ray spectroscopy (SEM/EDX).

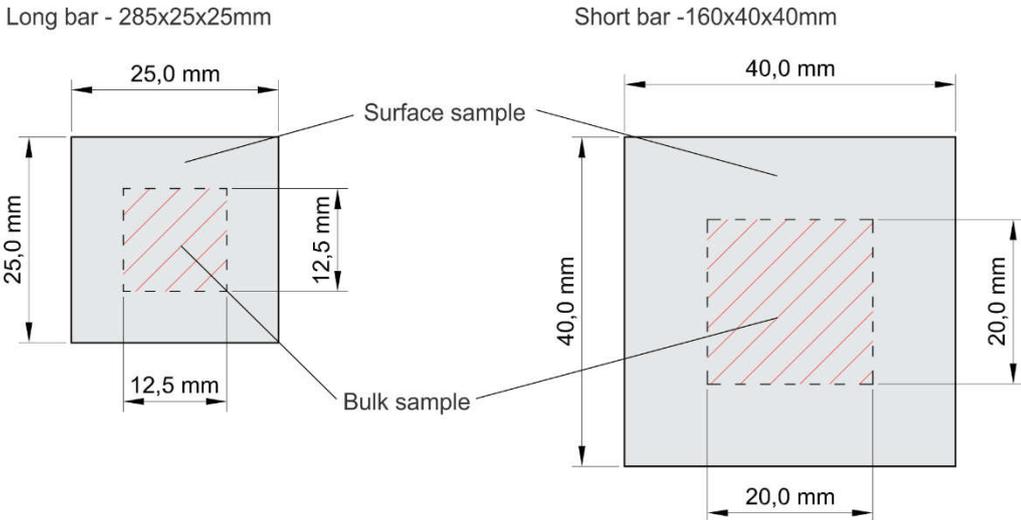


Figure 2 – Extraction of samples for TG analysis

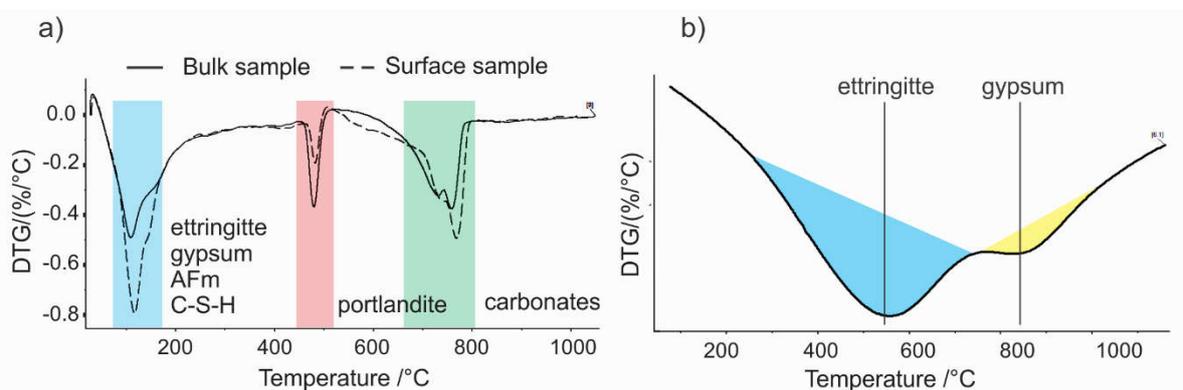
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**2.4. Thermogravimetric quantitative analysis**

Thermogravimetric analyses were performed on samples of representative long bars of the O, L, and H series for both expositions (solution and soil). Short bars of the H series were also analyzed by this method to observe the effect of size. Figure 3.a shows an example of DTG curves collected (bulk sample and surface sample) and the characteristic peaks analyzed. These peaks are related to the weight loss caused by dehydration, dehydroxylation, decarbonation, or decomposition of cement hydrates and ESA products. In the range measured, all samples showed peaks at similar locations due to the loss of weight. In the range of 50 to 200 °C, there is a significant overlap of peaks, including the decompositions of ettringite and gypsum (100 to 150 °C), hydrogarnet (around 140 °C), and AFm phases (multiple peaks between 60 and 200 °C). Moreover, C-S-H contributed to weight loss due to dehydration (65 to 600 °C, approximately). Later on, there are two isolated peaks corresponding to the

220 dehydroxylation of portlandite (400 and 480 °C) and the decomposition of carbonates (660 and  
 221 780 °C). The slight deviations from the temperature ranges normally reported for these phases  
 222 are explained by variations in the fineness and amount of the tested sample. The ESA process  
 223 is expected to consume portlandite to form gypsum and ettringite. Moreover, the carbonation  
 224 process also consumes portlandite to form carbonates. Calcite and portlandite contents might  
 225 also be affected by the FRA content, due to its attached cement paste.  
 226

227 The tangential method was used for the quantification of mineral phases, as described in [48].  
 228 This method involves measuring the area of the corresponding peak in the DTG curve. For this  
 229 purpose, the onset and the end of each peak were discerned from the abrupt change in the  
 230 curvature of the curves, and a straight line was drawn connecting these two-points, as shown  
 231 in Figure 3.b. The enclosed region was calculated by integrating the area beneath the secant  
 232 and subtracting the area below the DTG curve in the temperature domain. This quantifies the  
 233 loss of water or CO<sub>2</sub> (depending on the phase under analysis), and stoichiometric calculations  
 234 were used to estimate the total amount of each phase. This methodology reduces the error in  
 235 quantification due to substantial peak overlap in that the temperature range as opposed to the  
 236 computations based on the full weight loss step within a given temperature range. It primarily  
 237 allows for the decoupling of the phases of interest from the dehydration of C-S-H. It should be  
 238 noted that the quantifications were made for samples with a high content of aggregates, which  
 239 has the advantage of being more representative of concrete in service, but the disadvantage  
 240 of significant dilution of phases of interest. Regardless, the outcome achieved and  
 241 subsequently evaluated enables a beneficial comparative analysis among the various samples  
 242 in view of their corresponding composition.  
 243



244  
 245 Figure 3 – Peak identification and quantification procedure based on DTG curves: a)  
 246 examples of DTG curves and characteristic peaks, a bulk sample (solid line) and a surface  
 247 sample (dashed line); b) tangential method for the quantification.  
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249 The selection of tangent points varied across the samples, with some showing shifts in the  
 250 temperature range for the characteristic peaks due to methodological testing parameters such  
 251 as sample size and fineness. Typically, ettringite was quantified in the range of 90-135 °C,  
 252 corresponding to the highest first peak observed in this temperature range. The gypsum peaks  
 253 were substantially overlapped by other dominant peaks (i.e., ettringite, alumina phases, C-S-  
 254 H), hence the quantification (potentially underestimated) was considered within the visibly  
 255 identifiable peak range of 130-170 °C. Portlandite was quantified in well-defined peaks that  
 256 commonly fell within the range of 420-480 °C, and carbonates were primarily quantified within  
 257 the range of 600-800 °C.  
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### 3. MACROSCOPIC RESULTS

#### 3.1. Hardened properties

Table 4 presents the flow values in the fresh state, as well as the hardened properties evaluated at 28 days for mortars. An effect of the FRA content on flowability of mixes is observed. The high water absorption of FRA reduces the free water content, thereby affecting mortar mobility. Slightly lower flow values in the fresh state were obtained for H mortars, which can be explained by a greater water demand by the H cement due to its greater fineness compared to other cements. However, similar compaction degrees were achieved with the standard compaction procedure applied. Therefore, no impact on the hardened properties due to the consistency of mixes is noticed, i.e. comparable values for the density and water absorption.

Regarding these hardened properties, the main observed effect is from the FRA. As FRA increase, strength and density decrease, while water absorption increase. These results are in agreement with the literature [15,49]. The increase in porosity can be attributed to the inherent porosity of the FRA. Also, the lower flow values in FRA bearing mixes could induce porosity due to the presence of air-entrapped voids. However, the observed differences in flow values did not appear to significantly affect the compactness of the hardened mortars (compressive strength and density). This is possible since all mortars showed sufficient workability and were consistently compacted.

Table 4: Fresh and hardened properties of mortars (at 28 days).

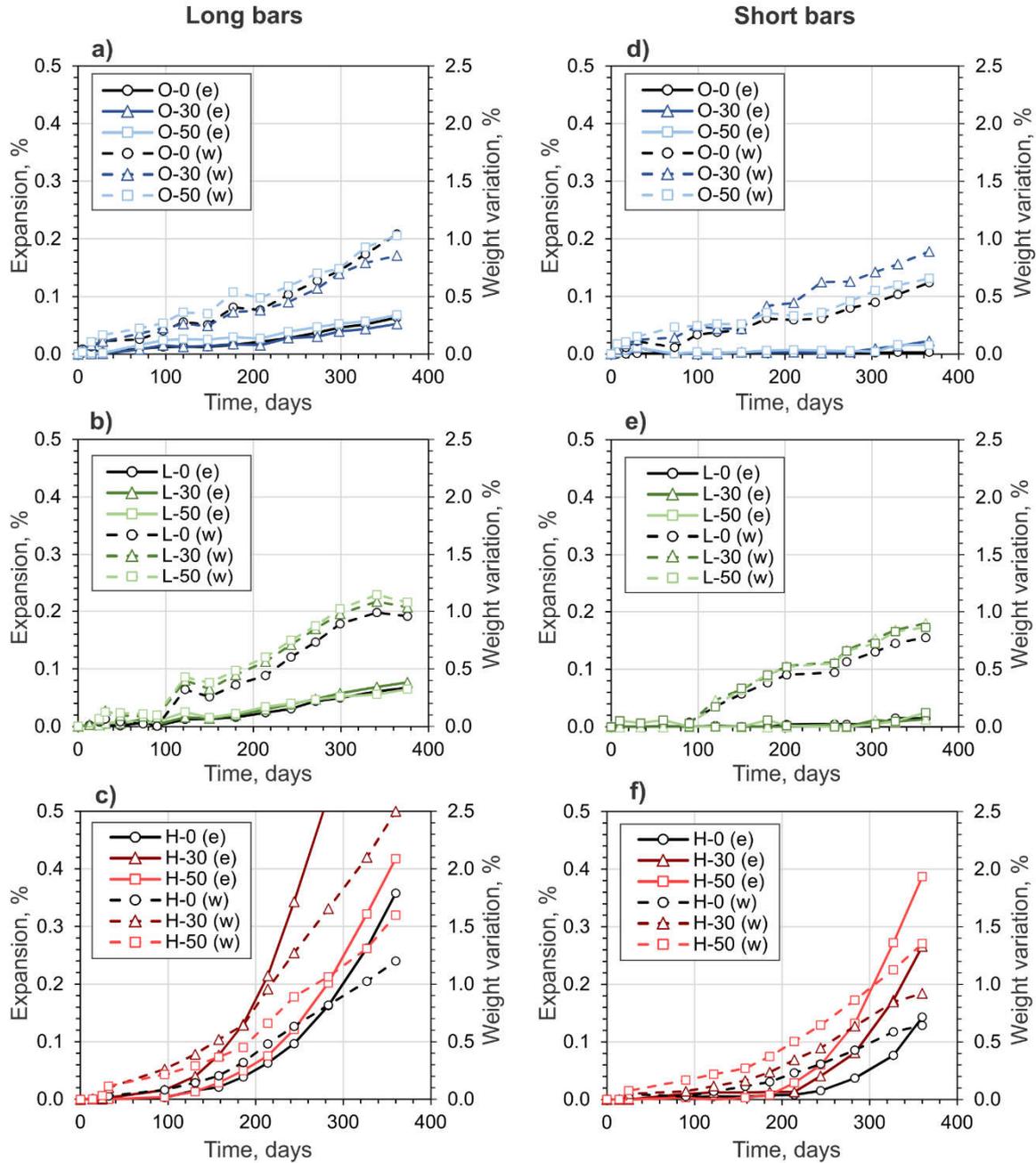
Mortar ID	Flow [%]	Flexural strength [MPa]	Compressive strength [MPa]	24h water absorption [%]	Density [kg/dm <sup>3</sup> ]
O-0	109	8.1	53.0	7.5	2.08
O-30	95	8.4	52.1	9.4	1.97
O-50	85	8.4	47.3	10.9	1.91
L-0	108	7.1	40.0	7.2	2.08
L-30	107	7.3	42.1	9.1	1.97
L-50	111	7.6	42.4	10.5	1.94
H-0	82	7.0	48.6	7.8	2.09
H-30	84	7.1	46.2	9.1	2.04
H-50	81	7.5	44.9	9.8	2.00

#### 3.2. Solution-exposed bars

Figure 4 shows the average expansions (solid lines) and weight changes (dashed lines) for each series of long and short mortar bars during 1 year of exposure to sulfate solution. The results show a main effect of the cement type of the recycled mortar on the final expansion of mortar bars and a secondary effect of the slenderness of the specimen. Only H mortars reached the propagation stage of ESA. FRA content shows a significant effect only for mortars with a low-performance matrix (H mortars).

For long bars, the L and O mortars had average expansions of less than 0.1%. This is the maximum limit after 1 year for compliance with ASTM C1012 [50] and is consistent with their

295 low-C<sub>3</sub>A cement. In contrast, H mortar bars had average expansions up to three times this  
 296 value (0.35, 0.85, and 0.42 % for 0, 30, and 50% FRA contents, respectively).  
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 298



299  
 300 Figure 4 – Expansion (solid line) and weight variation (dashed line) of mortar bars immersed  
 301 in sulfate solution: (a), (b), and (c), long bars of mortar series O, L and H, respectively; (d), (e),  
 302 and (f), short bars of mortar series O, L, and H, respectively.  
 303

304 For short bars, the average expansions in O and L mortars were less than 0.02% (effect of  
 305 slenderness), but for H mortar, the expansions were 0.12, 0.26, and 0.40%, for 0, 30, and 50%  
 306 FRA contents, respectively. In all cases, the average weight change showed a good correlation

307 with the average expansion, supporting the correlation between the precipitation of ESA  
308 products and expansion.

309  
310 Short bars had lower expansion values than long bars, as expected, but weight changes were  
311 similar for both bar types. This is due to the fact that both specimen types have approximately  
312 the same exposed surface area (260 cm<sup>2</sup> for long bars versus 224 cm<sup>2</sup> for short bars) but  
313 different cross-sectional areas. The expansion of mortar bars is correlated with the ratio  
314 between the sample volume under attack and the total volume of specimen [51], and this ratio  
315 decreases with decreasing slenderness. Moreover, a longer ESA induction period is observed  
316 for short bars compared to long bars for H mortars (~200 days for short bars and ~100 days  
317 for long bars).

318  
319 The results also show that the FRA content has an influence on mortar expansion only when  
320 the cement of the new matrix has a high C<sub>3</sub>A content (> 8 wt.%). However, the extent of this  
321 influence depended on the slenderness of the bars. For long bars, the H-30 mortar showed  
322 higher expansion than the H-50 mortar, while for short bars, the H-50 mortar showed a higher  
323 expansion than the H-30 mortar. The weight changes are consistent with these trends. The  
324 results suggest that the cross section of the specimen changes the pessimum value for the  
325 FRA content in mortar under ESA. This ambiguous effect of recycled aggregates on mortars  
326 performance has been earlier reported in the literature [24,31,52], although it seems to be  
327 influenced by other variables of the experimental set-up (see Section 5 for more details).

### 328 329 **3.3. Soil-exposed bars**

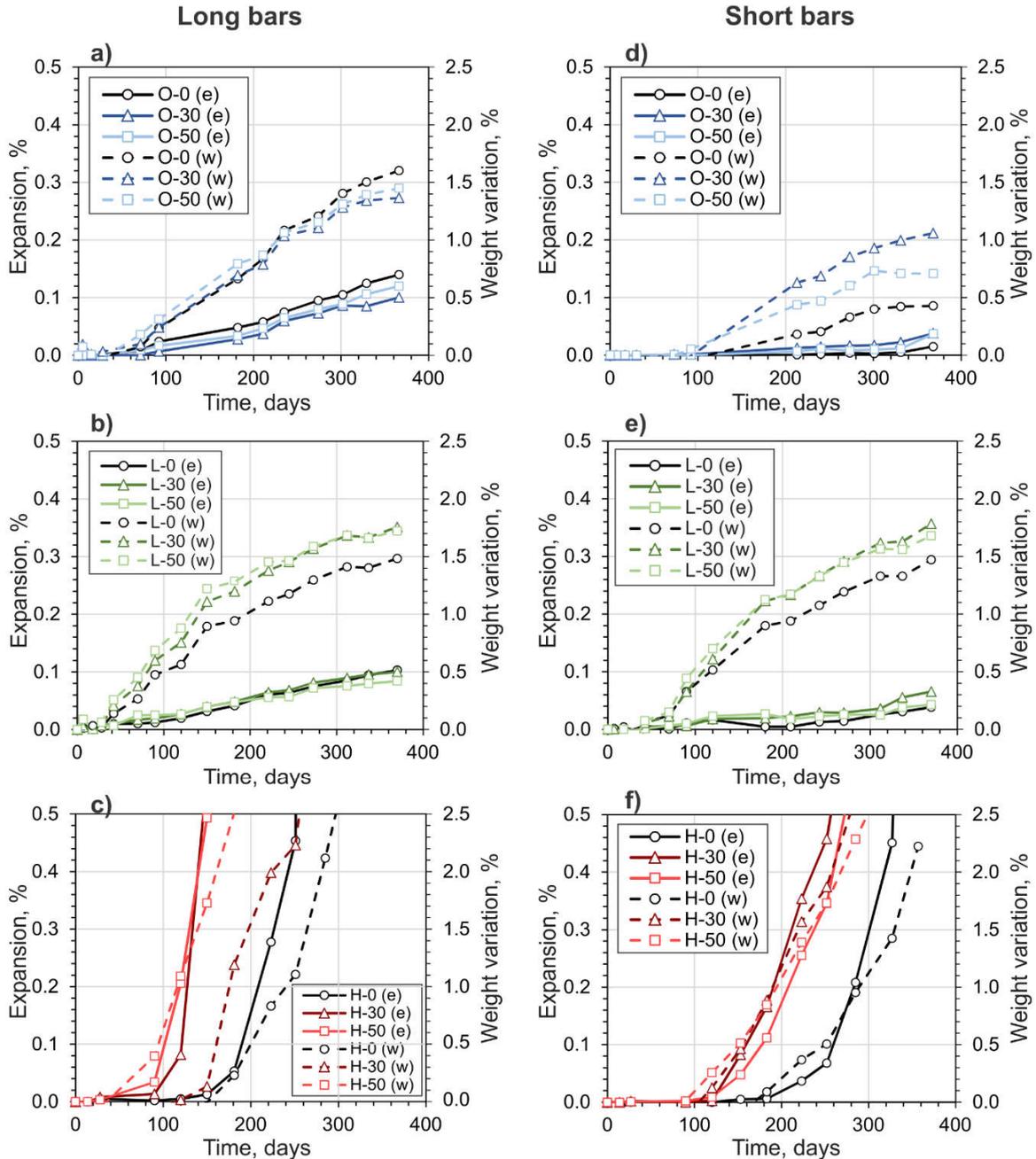
330  
331 Figure 5 shows the corresponding results of average expansions (solid lines) and weight  
332 changes (dashed lines) for mortar bars buried in sulfate-rich soil. All values  
333 increased compared to the corresponding specimens in sulfate solution, which can be  
334 attributed to the effect of PSA. As with the sulfate solution, the cement type was critical to the  
335 behavior of the mortars, and only the mortars of the H series showed a propagation of ESA.  
336 Also, effects of slenderness and FRA content were observed in a second order. There is a  
337 good correlation between the expansion and the weight change, confirming the process of  
338 expansion by precipitation of ESA products inside the specimens.

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340 For long bars, values above 0.1 % (limit criteria of ASTM C1012) are observed after 1 year of  
341 exposure, even for mortars made with low-C<sub>3</sub>A cements. This can be attributed to the  
342 development of PSA, which is a physical attack and does not depend on the C<sub>3</sub>A content of  
343 the cement. However, the H mortars still suffered significantly greater expansion than the O  
344 and L mortars, confirming a combination of salt crystallization and ESA that may produce a  
345 synergistic attack mechanism. In fact, the deterioration of the specimens was quite severe,  
346 and some specimens of the H mortars were destroyed even before 1 year of exposure was  
347 completed.

348  
349 The size of the specimens played a similar role to the sulfate-solution exposure. For long bars,  
350 the O and L mortar bars expanded about 0.1% after 1 year of exposure, whereas for short  
351 bars, expansions were lower than 0.05%. For the O mortars, short bars showed a lower  
352 expansion and weight change than long bars. For the L mortars, short bars expanded less but  
353 showed a similar weight change to that of long bars. For H mortars, a small reduction in the  
354 propagation rate was observed in short bars, but the high damage observed for both sizes  
355 makes this difference irrelevant.

356

357 In this case, slight and inconsistent effects of FRA are observed in low- $C_3A$  cement mortars  
 358 and are attributed to its effect on porosity. For long bars, the O and L mortars with FRA showed  
 359 the same or slightly lower expansions, while for short bars the effect was reversed. Weight  
 360 variations were also higher for short bars when FRA was used, except for long bars made with  
 361 O mortar, where the opposite effect occurred. In H mortars, the FRA content changed the  
 362 duration of the induction period ( $\sim 100$  days for H-30 and H-50, and  $\sim 150$  days for H-0).  
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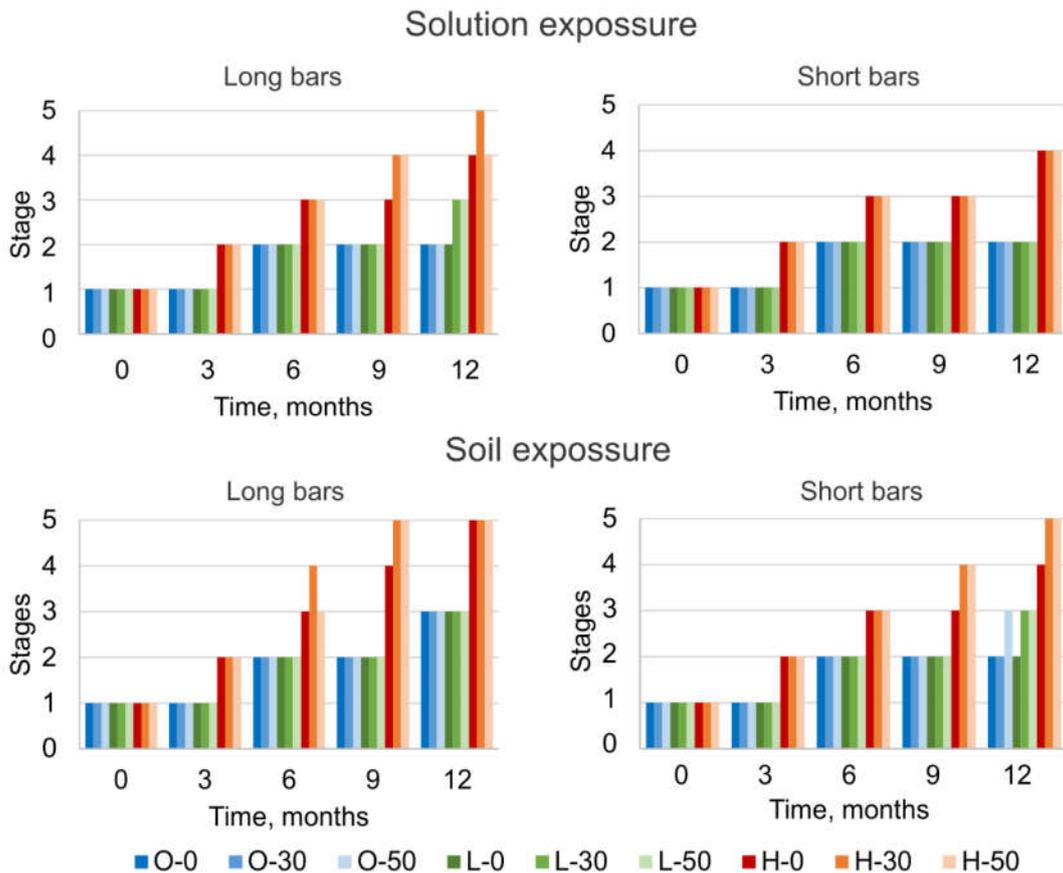


364  
 365 Figure 5 – Expansion (solid line) and weight variation (dashed line) for mortar bars buried in  
 366 sulfate soil: (a), (b), and (c), long bars of the O, L, and H mortar series, respectively; (d), (e),  
 367 and (f), short bars of the O, L, and H mortar series, respectively.  
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369 When analyzing the results from sulfate soil exposure, it should be mentioned that the standard  
 370 deviations were higher than when exposed to sulfate solution (they were not plotted to maintain  
 371 readability in the graph). The level of damage is more variable, which is probably due to local  
 372 defects in the mortar bars that allow PSA to open cracks faster and trigger the expansion  
 373 period. It can be summarized that the content of FRA was only significant in a poor chemical  
 374 quality cement matrix, reducing the sulfate resistance of mixes. No results on mortars with  
 375 recycled aggregates under these exposure conditions were not found in the literature.  
 376 However, these findings align with results of concretes in comparable exposure conditions  
 377 [30,31].

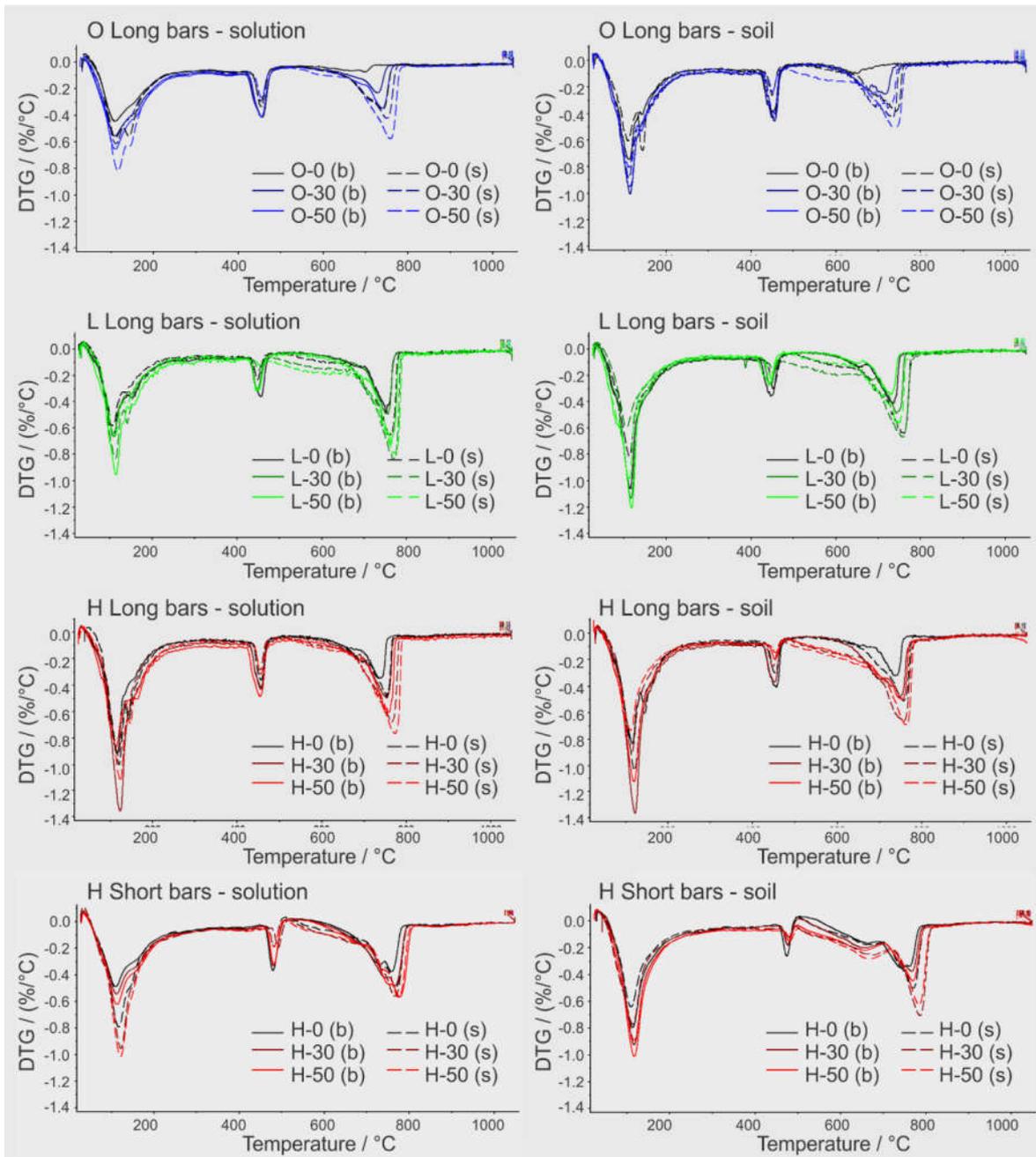
### 379 3.4. Visual inspection of damage

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 381 The results of the visual inspection are presented in Figure 6. There is a strong correlation  
 382 between the length and weight changes and the progression of deterioration stages. On the  
 383 one hand, long bars showed a higher degradation than short bars for the same exposure type.  
 384 On the other hand, same size specimens were more damaged when exposed to the sulfate  
 385 soil (ESA+PSA). Only H mortars reached stages 4 and 5 of damage. The time to reach stage  
 386 4 for each series had a good correlation with the time of the induction period observed in  
 387 expansion and weight variation curves. Finally, a slight increment of cracking development is  
 388 observed in soil-exposed samples of series O and L when they have a FRA replacement. This  
 389 difference can be attributed to the higher porosity of FRA. This increased porosity results in  
 390 higher transport properties, which can accelerate the PSA action, regardless of the cement  
 391 type of the new matrix.



392  
 393 Figure 6 – Deterioration stages by visual inspection, up to 12 months.

394 In qualitative terms, it was observed a slight difference in the deterioration mechanisms of  
 395 specimens regarding the exposure type. Solution-exposed bars showed progression a bit more  
 396 uniform than soil-exposed bars. On the one hand, exposure to sulfate solution is based mainly  
 397 on diffusion, which is uniform throughout the whole bar. The cracking development was uniform  
 398 as described by the stages, and a high bending of the shaft was observed in all samples before  
 399 failure. On the other hand, in the soil-exposed specimens, there is also a sulfate ingress with  
 400 water transport. This mechanism is more sensitive to local defects (big pores, mesocracking,  
 401 etc.). In this case, the cracking development was more concentrated in spots, and the failure  
 402 of specimens was earlier than shaft bending in most of cases  
 403



404 Figure 7 – DTG curves of mortars after 1 year of exposure  
 405

## 4. MICROSTRUCTURAL ANALYSIS

### 4.1. Thermogravimetric analyses

Figure 7 shows the DTG curves for all samples analyzed. Although all samples showed similar peak distributions (characteristic peaks described in the Methodology section), some differences are observed owing to the location of the sample (bulk or surface), as well as the FRA content or the cement type of the mortars.

In general terms, surface samples showed lower contents of portlandite with respect to bulk samples. Moreover, all mortars showed a higher content of carbonates at the surface than in the bulk. Both phenomena are attributed to the carbonation of samples during preparation and the exposure of samples, which has a gradient from the surface to the bulk. The same is observed for long bars exposed to sulfate soil, with slight differences on the carbonates' peak shapes. In both portlandite and carbonates peaks, there is also a visible effect of FRA content. The higher the FRA content, the larger the carbonate peak. However, it should be kept in mind that initial carbonate content might be affected by FRA content as it contains attached (probably carbonated) cement paste (continue in Section 4.1.1).

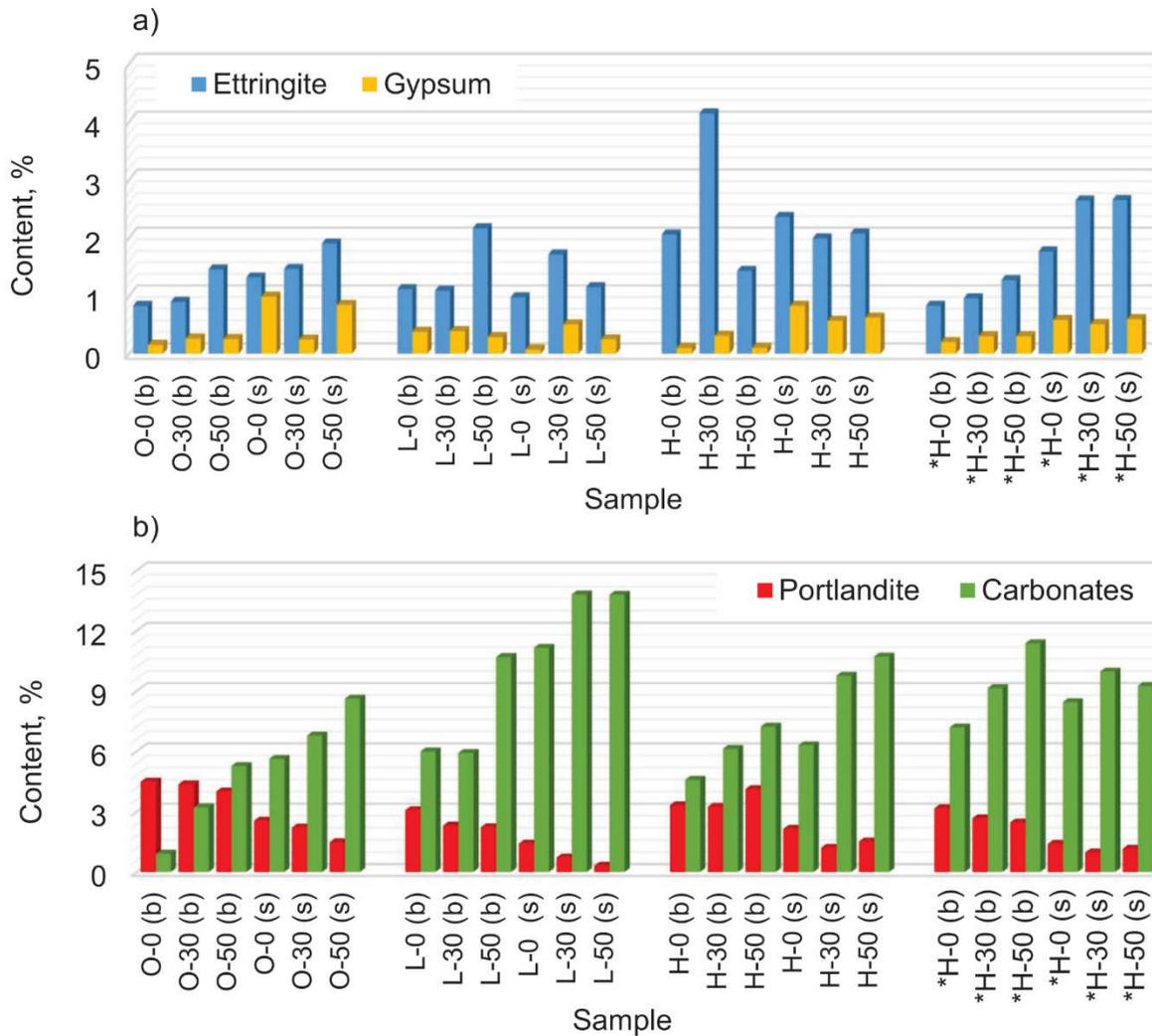
Regarding the range of 50-200 °C, there are clear differences regarding the cement type. For long bars exposed to solution, H mortars showed bigger peaks than L and O mortars, even in bulk samples. Bigger peaks in this range are observed for series exposed to sulfate soil, which evidences a higher extent of deterioration for ESA in combination with PSA. H-mortar short bars showed smaller peaks than long bars in this range, especially in bulk samples. This is attributed to the size effect on the ESA kinetics.

#### 4.1.1. Solution-exposed samples

Results of mineral quantifications (calculated as mentioned in section 2.4) for solution-exposed samples are presented in Figure 8. In general, mineral contents differed consistently between bulk and surface samples, consistent with the mechanisms of ESA that start at the surface and progress in the bulk. It was also observed an effect of the cement type and the FRA content in mixes. When comparing the last two groups (long and short bars of the H series), some differences linked to the slenderness of specimens are also observed. Relative amounts of cement phases are presented relative to the total weight of the samples. These values are comparable to mineral quantifications of cement-based materials affected by ESA reported in the literature [33,53].

In terms of ESA products, there was a slightly higher presence of ettringite and gypsum observed in the H mortars compared to the O and L series. Additionally, the H-30 mortar, which had exhibited the highest level of macroscopic deterioration, showed also a greater amount of ettringite in its bulk sample. It's important to note that the total quantity of ESA products does not necessarily correlate with the extent of deterioration, since damage occurs when the material's buffer capacity is depleted [33,53], and according to the crystallization pressure hypothesis, only ettringite present in small micropores is responsible for initiating cracking. However, in this case, the non-linear impact of FRA content on ESA progression was also observed in thermogravimetric quantification. The positive effect of a higher buffer capacity and the negative effect of a higher transport rate, make ambiguous the effect of the FRA content.

456 Ettringite contents are always higher in surface samples than in bulk samples (except in H-30  
 457 mortar), and the same occurs with gypsum. As sulfate ions come in from the surface, both  
 458 minerals are found in greater quantities in surface samples. Higher contents of gypsum were  
 459 observed on the surface than in the bulk for almost all series. Because the amounts of gypsum  
 460 determined are quite small, the quantification accuracy might be lower (near the detection limit  
 461 of the method). For H-series short bars, the content of ettringite in bulk samples is lower than  
 462 that in long bars, and even similar to that in O mortars. However, higher contents of ettringite  
 463 are observed in surface samples. This evidences a different mechanism of deterioration by  
 464 ESA depending on specimen size and shape.  
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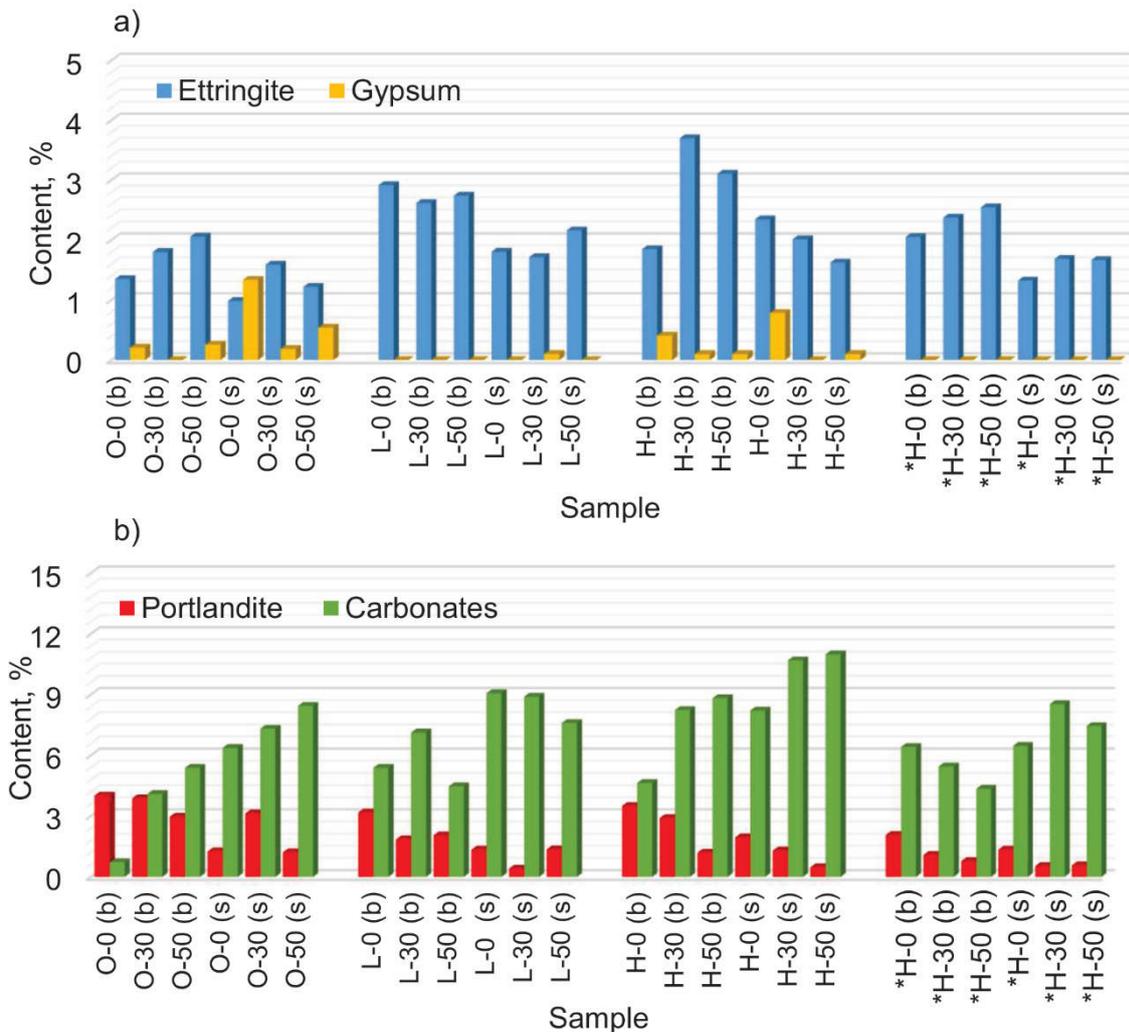
467  
 468 Figure 8 – Mineral quantification in mortar bars exposed to sulfate solution for up to 12  
 469 months: a) ESA products; b) common cement paste minerals.  
 470 (Mixes nomenclature: b=bulk samples; s=surface samples, \*=short bars)  
 471

472 The reduction of portlandite amounts in samples is related to the degree of carbonation, which  
 473 consumes it to form calcite, and to the development of ESA, which consumes it to form gypsum  
 474 and ettringite. In all samples, more portlandite was found in the bulk than in the surface  
 475 samples. Then, for all groups, a higher portlandite loss is observed when FRA content

476 increases. The only exception to this is for H mortar long bars. In this case, the lowest  
 477 portlandite content corresponds with H-30 mortar (higher ettringite+gypsum content and higher  
 478 expansion). Initial contents of carbonates are linked with the cement type of the mortar  
 479 (limestone filler) and can be affected by the attached mortar of FRA. Assuming that all attached  
 480 mortar of FRA was carbonated before used, the maximum increase in total carbonate content  
 481 could be up to approximately 2 and 4% for 30 and 50% FRA content, respectively. Then,  
 482 results showed higher increments of carbonate contents in samples when FRA content  
 483 increased; hence, the higher porosity of mortars with FRA might have accelerated the  
 484 carbonation progression.

485  
 486 **4.1.2. Soil-exposed samples**  
 487

488 Results of mineral quantifications for soil exposure samples are presented in Figure 9. As in  
 489 solution-exposed samples, there are clear differences between bulk and surface samples,  
 490 which are related to the deterioration mechanism. Then, there are observed differences  
 491 regarding the cement type and FRA content of mortars, and the slenderness of specimens.  
 492 Moreover, the PSA action might increase the permeability of specimens to aggressive agents,  
 493 changing the combined mechanism of deterioration.  
 494



495

496 Figure 9 – Mineral quantification in mortar bars exposed to sulfate soil for up to 12 months: a)  
497 ESA products; b) common cement paste minerals.  
498 (Mixes nomenclature: b= bulk samples; s= surface samples, \*= short bars)  
499

500 In contrast with solution-exposed samples, higher contents of ettringite are observed in bulk  
501 samples than in surface samples, in general. In this case, the PSA mechanism accelerates the  
502 ion ingress to the whole sample, and later, it seems that ESA post-mortem products were more  
503 stable in bulk samples than in surface samples. Furthermore, the effect of the cement type is  
504 not as visible as in the solution-exposed series. For instance, the L series seems to show  
505 similar ettringite contents to the H series. It should be noted that H mortars had suffered a  
506 severe degradation at the moment of sample extraction, whereas L specimens had shown  
507 small and isolated cracks, similar to the O series. This may allow a loss of post-mortem ESA  
508 products during the manipulation of the specimens (they were washed after each  
509 measurement) and samples extraction.

510  
511 For example, the control mortar of the L series showed a higher ettringite content than the H  
512 control mortar, but the latter showed a gypsum content, which is a post-mortem ettringite  
513 remainder. Also, gypsum was detected in surface samples of O mortars, which had a higher  
514 relative content of portlandite than limestone filler types of cement (L and H).

515  
516 Regarding the FRA content, the ettringite amount showed different trends according to the  
517 cement type. For the O series bulk samples, a higher ettringite content is observed as FRA  
518 content increases, but for surface samples, a higher content is observed for 30% FRA mortar.  
519 For L series, no significant variations due to FRA contents are observed.

520  
521 For the H series, mortars with FRA showed more ettringite content in the bulk than the control  
522 mortar. However, surface samples showed lower contents of ettringite as FRA content  
523 increased. This is showing that considerable amounts of ettringite could have precipitated as  
524 gypsum in an advanced stage of attack, and then some of the latter was lost during the sample  
525 extraction and preparation processes.

526  
527 Portlandite contents showed, in general, a reduction when FRA content increased, and,  
528 secondarily, a reduction in surface samples compared to bulk samples. Both phenomena are  
529 related with the ESA and carbonation mechanisms. Results suggest that the higher porosity  
530 of FRA increased the kinetics of both phenomena. However, there are some samples that do  
531 not comply with the general trend. For the O series, a clear relationship is observed between  
532 the portlandite reduction and gypsum observed in surface samples. For the L series, a lower  
533 portlandite content is observed in the mortar with 30% FRA.

534  
535 Regarding the carbonates, there are in general higher contents in the surface samples than in  
536 those from the bulk, but variable effects of the FRA content for each series. For O and H, long  
537 bar series, the carbonate content was quite proportional to the FRA content. Conversely, for L  
538 mortars, the general correlation was in the opposite direction, as in the bulk samples of H-  
539 mortar short bars. Then, higher amounts of carbonates were found in mortars with 30% FRA.  
540 Thereby, it seems that the more heterogeneous cracking mechanism of the ESA+PSA  
541 exposure condition might induce difficulties in the sample extraction and its representativeness  
542 at this scale of analysis.

543  
544 The earliest damage observed in H mortars is likely related to higher ettringite formation due  
545 to the high C<sub>3</sub>A content in the cement. It is possible that a certain amount of ettringite

546 decomposes when porosity suddenly increases and thermodynamic conditions change due to  
547 cracking. Normally, ettringite decomposes to gypsum when the pH decreases or the sulfate  
548 concentration is high [36,54,55]. Thereafter, it is possible that limestone cements form  
549 monocarbonate, which is a more stable phase than gypsum [55] and whose peak is near the  
550 range of gypsum and ettringite. This could result in an overestimation of ettringite contents in  
551 L mortars, for instance.

552

## 553 4.2. Microscopic analyses

554

555 Microscopic and compositional analyses were performed using SEM/EDX instruments.  
556 Microscopic inspection was done on samples taken from the same specimens as for  
557 thermogravimetric analyses. Analysis included general characterization of cracking,  
558 examination of ESA-related compounds, and interaction of FRA with attack progression  
559 (analysis of surface and bulk zones of specimens).

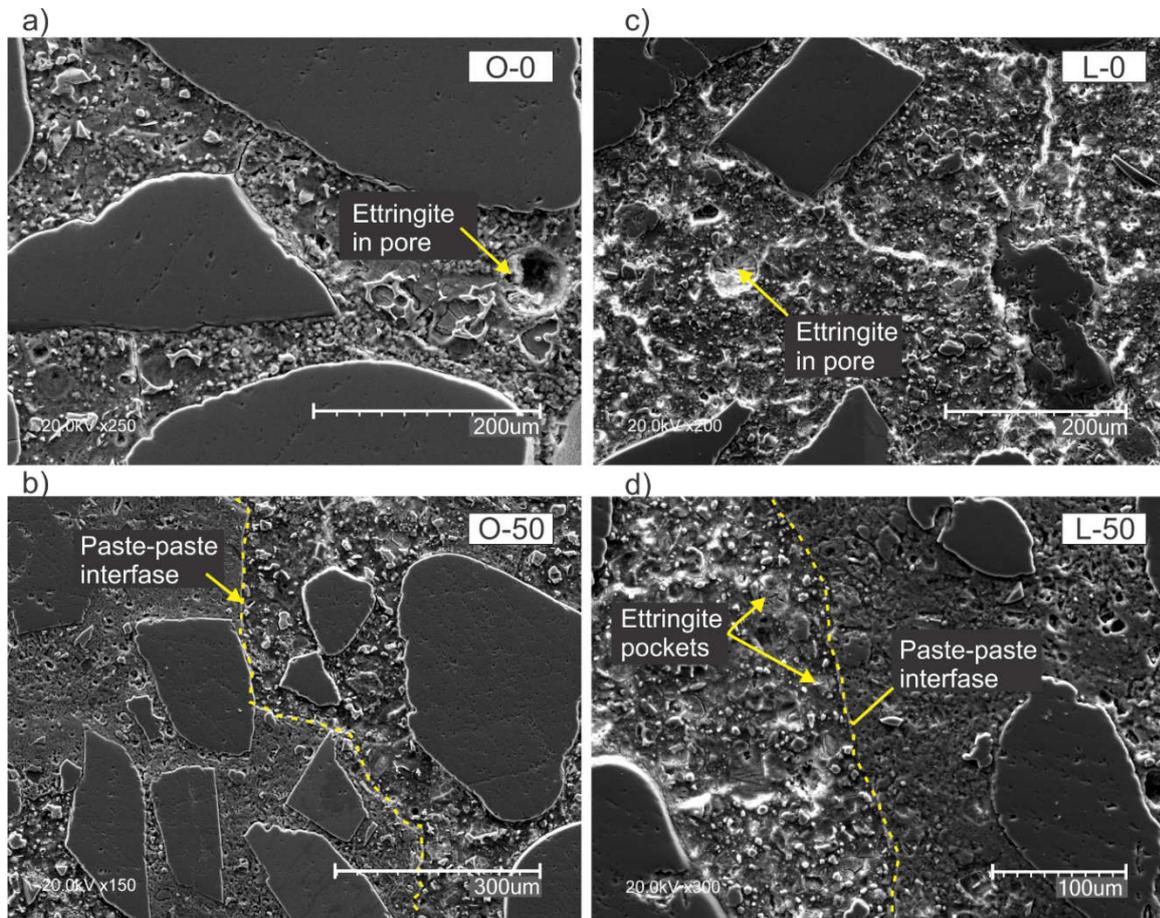
560

### 561 4.2.1. Solution-exposed samples

562

563 Microscopic inspections revealed signs of ESA due to the presence of ettringite and gypsum  
564 in all samples, even in the mortar samples with low- $C_3A$  cement. Also, different levels of  
565 cracking were observed in mortar samples. Special attention was given to RCA particles and  
566 their surroundings, in order to know their participation in the ESA process.

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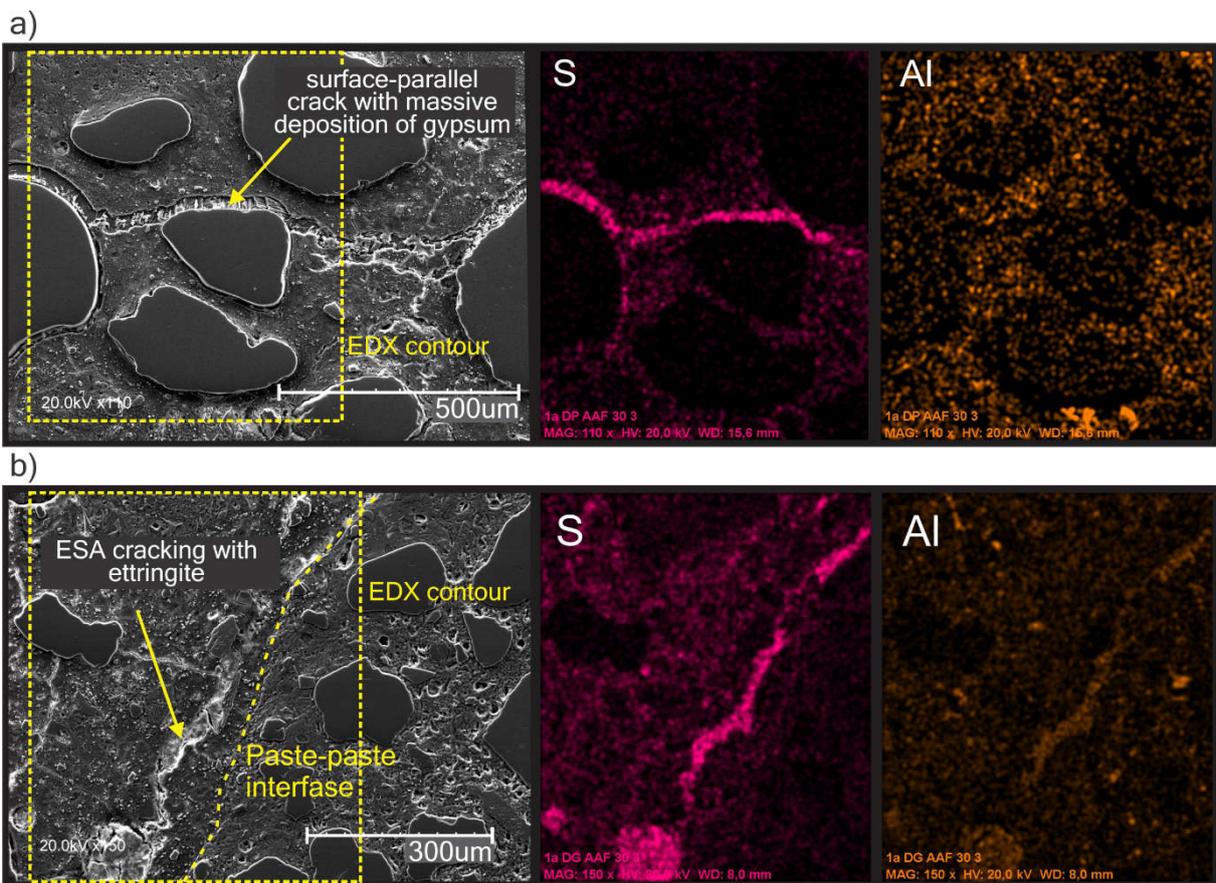


568

569 Figure 10 – SEM/BSE images of O and L mortar specimens after 1 year to sulfate-solution  
 570 exposure: a) O-0 sample; b) O-50 sample; c) L-0 sample; d) L-50 sample.  
 571

572 For samples of O and L mortars, some isolated microcracking (<5µm) was observed (Figure  
 573 10). However, since they were empty of ESA products in general, it is possible that they are  
 574 related to cracks induced in the sample extraction. In all samples, several small pores (between  
 575 30 and 200 µm) were observed fully or partially filled with ettringite. These entrapped-air voids  
 576 could act as a buffer for the mortars [33,56], altering the relationship between the ESA mass  
 577 changes and the rate of deterioration. As the use of FRA could increase air incorporation  
 578 [32,57], this could result in a positive effect of its use. Furthermore, ettringite was detected in  
 579 small pockets within the cement matrix, associated with production of non-connected cracking.  
 580

581 In the case of H mortar samples, the development of cracking was greater than in the O and  
 582 L series. The small pores of entrapped air were in general filled with ESA products, indicating  
 583 a consumption of the buffer capacity, and pockets of ettringite in the cement matrix were  
 584 observed with associated cracking. Moreover, for this series, a larger development of cracking  
 585 was observed, generally filled with ESA products (Figure 11). This is consistent with  
 586 macroproperties observed, where H mortars exhibited the highest deterioration values.  
 587

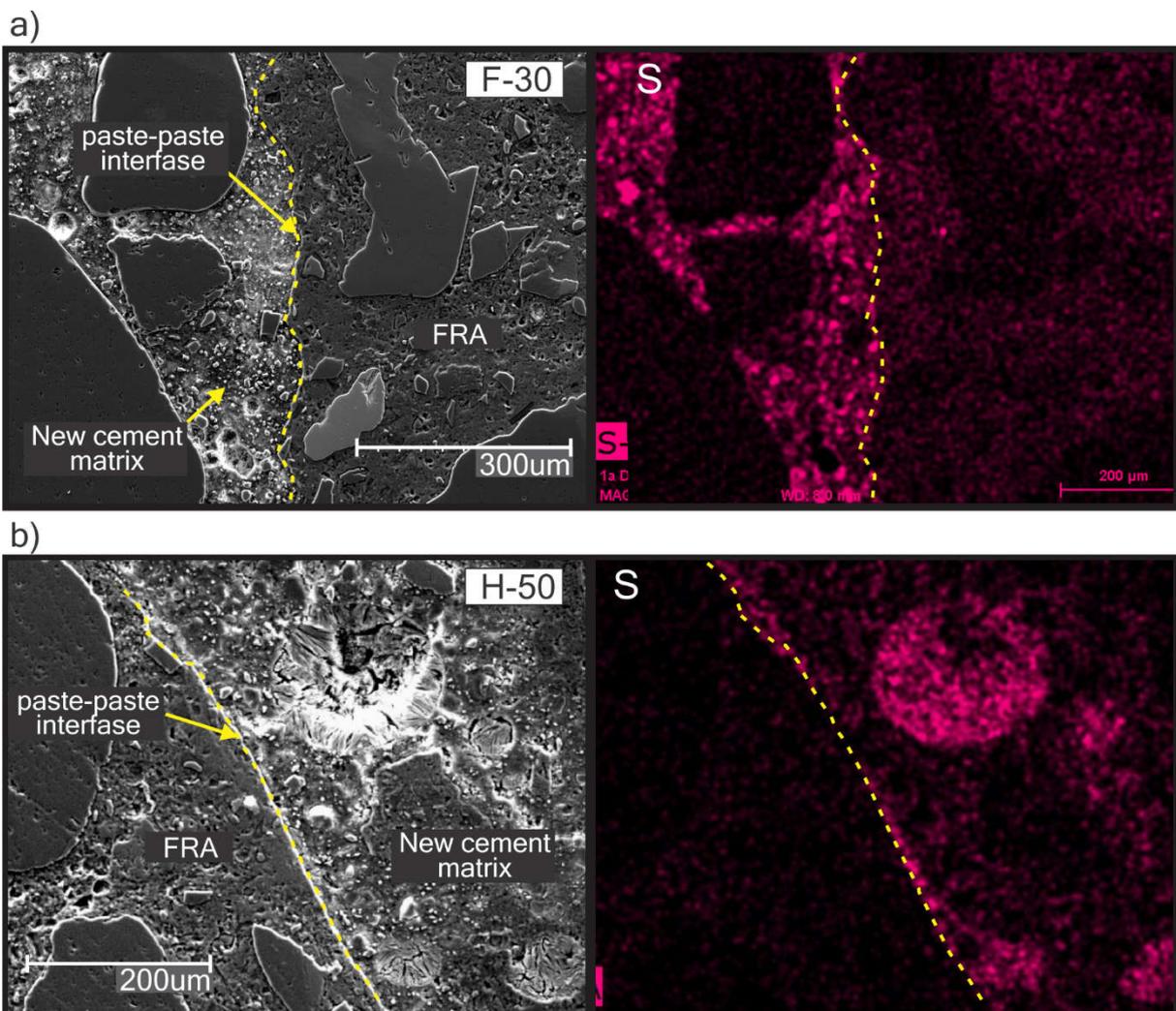


588 Figure 11 – SEM/BSE images and corresponding EDX mappings of sulfur (S) and aluminum  
 589 (Al); a) Near-surface region of an H-30 long bar; b) Bulk region of an H-30 short bar.  
 590

591 The image in Figure 11a was taken near the surface region of an H-30 long bar sample. A  
 592 crack of approximately 30 µm, filled, parallel to the surface, is observed, which opens through  
 593

594 the interfacial transition zone and the cement paste matrix. The EDX analysis confirms that the  
 595 filling material in the cracks corresponds mainly to gypsum (high sulfate concentration and no  
 596 aluminum). These types of filled cracks are post-mortem evidence of ESA propagation. The  
 597 higher crack formation near the surface allows the sulfate concentration to increase, and  
 598 gypsum becomes more stable than ettringite.  
 599

600 The image in Figure 11b was taken from the bulk region of an H-30 short bar sample. In this  
 601 case, the wider cracks related to ESA progression appear filled with ettringite since EDX  
 602 mappings showed a higher sulfur and aluminum concentration in there. In the same image, an  
 603 interface between the old mortar attached to an FRA particle and the new cement matrix is  
 604 observed. There, the ESA crack connects a part of the paste-paste interface with a pore of the  
 605 cement matrix fully filled with ettringite. This distribution of ettringite and gypsum in the bulk  
 606 and the surface, respectively, was generally observed, supporting the hypothesis that ettringite  
 607 is more associated with the formation of cracks (front of attack), and the presence of massive  
 608 gypsum serves as post-mortem evidence in larger cracks [34,58].  
 609



610  
 611 Figure 12 – SEM/BSE-EDX mappings of sulfur (S) images of interfacial transition zone  
 612 between a FRA particles and the new matrix of mortar, of samples exposed to sulfate  
 613 solution. a) L-30 mortar sample; b) H-50 mortar sample.

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Regarding the FRA inspection, it was generally observed that the attached cement paste had a lower sulfate content than the new cement matrix. Figure 12 shows two examples of the interfacial transition zone between a FRA particle and the cement matrix of the mortar (new and old cement paste). The upper image corresponds to a low  $C_3A$  cement mortar (L-30), and the lower image corresponds to a high  $C_3A$  cement mortar (H-50). In both cases, a higher concentration of sulfates in the new cement matrix than in the old attached cement paste is observed. This could indicate two possibilities: that the sulfate ions penetrate the FRA particles in lower amounts (incoherent porosity) or that the ions penetrated the FRA particles but did not found binding minerals to react with and precipitate (reduced AFm and/or portlandite content). It should be kept in mind that FRA cement paste probably has a high degree of carbonation, which means a lower content of portlandite, from which gypsum and then ettringite are formed. The microcracks associated with the precipitation of ESA products almost always occurred in the new cement matrix and not in the old cement paste. This situation was observed in all samples exposed to sulfate solution and confirmed by EDX point analyses.

#### 4.2.2. Soil-exposed samples

Samples exposed to sulfate-rich soil were found to develop a greater extent and width of crack compared to solution-exposed samples. This is attributed to the physical action of salt crystallization in combination with ESA.

In contrast with solution-exposed samples, in O and L samples, a small ( $< 5 \mu\text{m}$ ) but more notorious mapping-arrangement cracking was observed, especially near the surface. Figure 13 shows examples of this cracking development. The elemental composition did not show ESA products on these cracks, and this suggests that this cracking is due to the physical action of salt crystallization. They also could be formed by the samples' extraction, but the good state of these mortars bars reduces this possibility. As was observed in solution-exposed samples, it is observed that pores are fully or partially filled with ettringite.

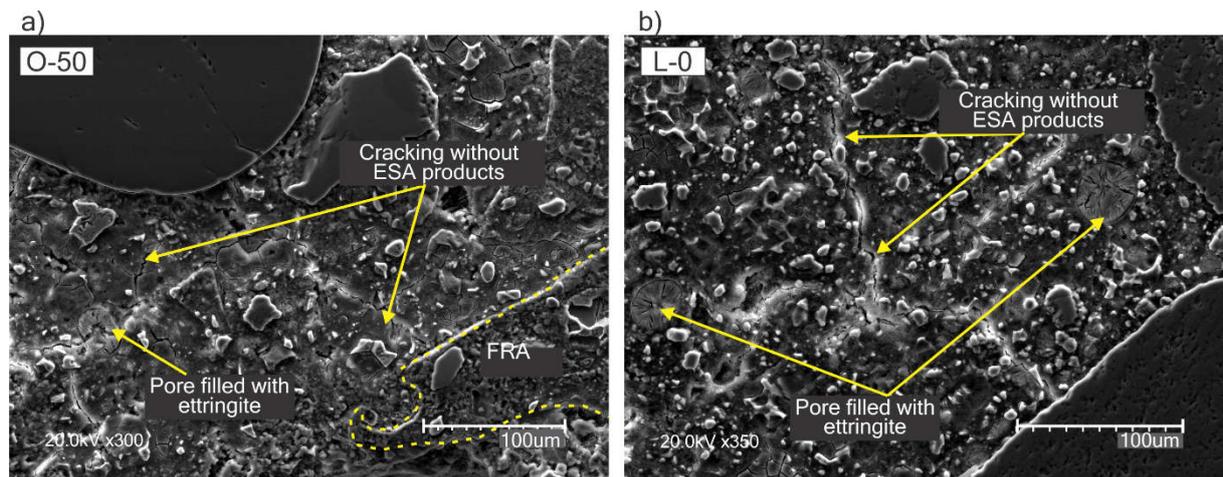
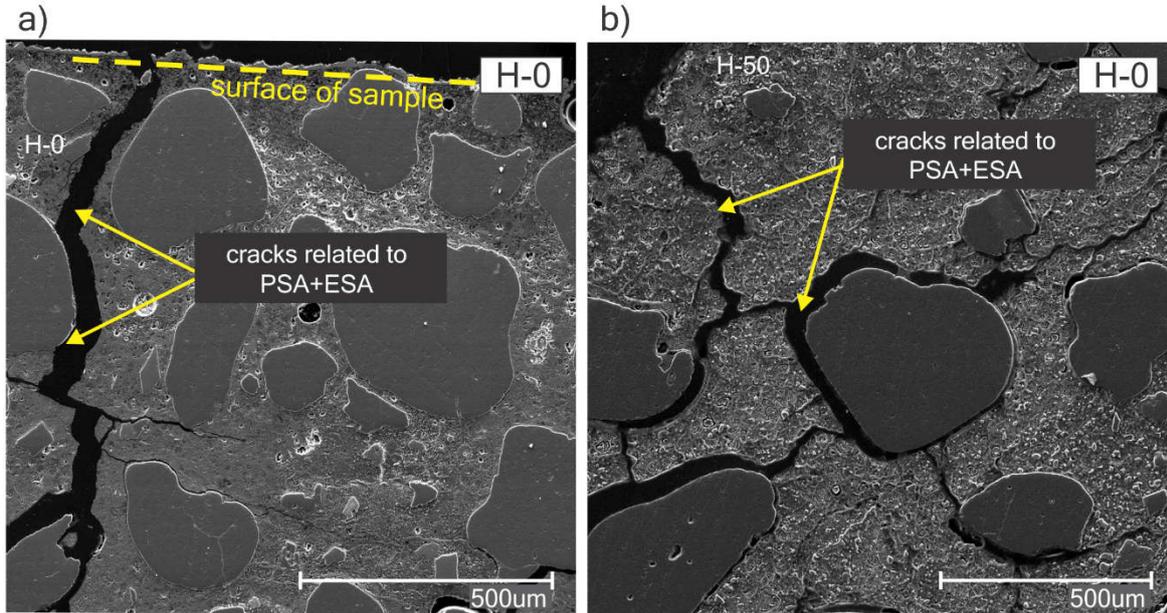


Figure 13 – SEM/BSE images of O and L mortar specimens after 1 year to sulfate-soil exposure: a) O-50 sample; b) L-0 sample.

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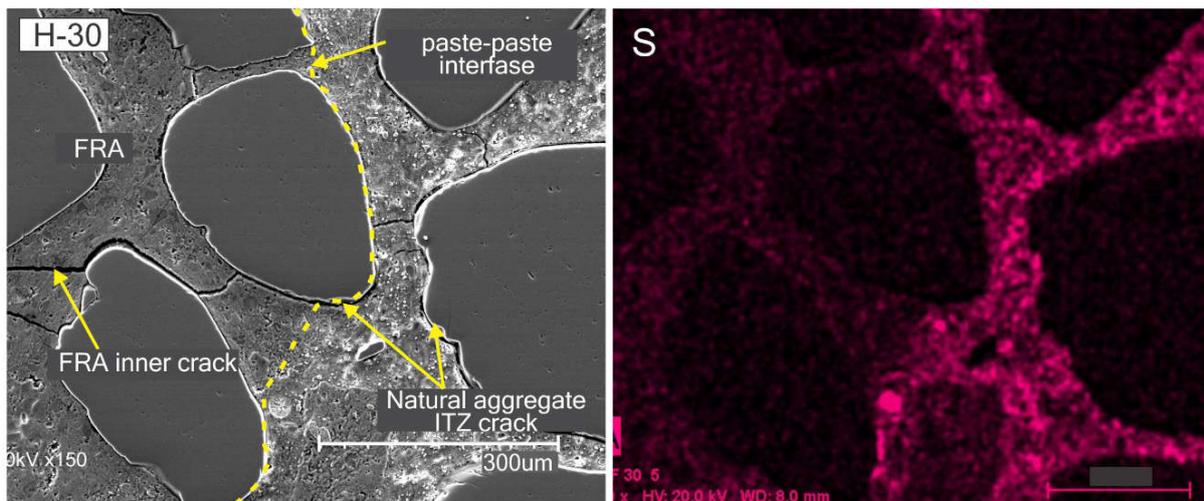
For H mortars the cracking development was greater, and also a larger-scale cracking is observed. In Figure 14, some examples of these cracks are presented. Cracks with even 100  $\mu\text{m}$  width are observed throughout the section, around natural aggregate interfaces and

651 cement matrix. A characteristic of these cracks is that they do not run parallel to the surface  
 652 as pure ESA cracks do. In some cases, they run perpendicular to the surface and penetrate  
 653 the sample with a pattern arrangement. Also, lower-scale microcracking is observed on the  
 654 cement matrix, with ESA products filling cracks or in small pockets.



655  
 656 Figure 14 –SEM/BSE images of specimen exposed to sulfate-rich soil up to 12 months: a)  
 657 Near surface region of H-0 mortar sample; b) Bulk region of H-0 mortar sample.  
 658

659 As was observed in samples exposed to sulfate solution, the sulfur mapping showed lower  
 660 binding of sulfate ions in the old cement paste attached to FRA particles (Figure 15). In general,  
 661 the attached cement paste from FRA showed a lower microcracking development than the  
 662 new cement matrix. However, it can be observed that some cracking development is coherent  
 663 between the matrix and FRA particles, as in the example presented in Figure 15. Also in this  
 664 case, the sulfur content in FRA paste is lower than in cement matrix.  
 665



666  
 667 Figure 15 - SEM/BSE-EDX images of interfacial transition zone between an FRA particles and  
 668 the new matrix of mortar H-30, exposed to sulfate soil.

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## 5. DISCUSSION

The results of these experiences shed light on the role of recycled aggregate particles in terms of sulfate attack and associated degradation processes (PSA and carbonation). The evaluation of macro-properties has shown that mortars with low alumina cement maintain their sulfate resistance even in mortars with 50% FRA content. In this case, the aluminum content of the source concrete from which the FRA was obtained is unknown, but it can be expected that it was made with a low- $C_3A$  cement, since Argentinian cement has a low content of this phase in general. When mortars were exposed to unsaturated conditions and PSA could develop, the higher porosity of mortars with FRA appears to have had a negative effect, even when using advantageous cement. Mortars with O and L cements (4.9 and 3.2%  $C_3A$  content, respectively) showed different behavior depending on the FRA content, but differences were not significant.

An FRA effect on mortars behavior was observed in the high  $C_3A$  cement used (H series mortar). Particularly in the case of solution-exposed samples, the mortars with 30% FRA showed worse performance than the control mortars and the mortars with 50% FRA. Although a high porosity of mortar (or concrete) is considered to enhance ESA kinetics and expansion, porosity affects through multiple mechanisms other than just increasing the sulfate penetration rate [33]. The porosity corresponding to the entrapped air may act as a buffer against the ESA products precipitation. The higher porosity of FRA also results in a lower stiffness of whole material, which can contribute to a reduction in the internal restraint of the specimen and less internal stresses [59], and consequently less cracking.

Microstructural inspection showed no significant precipitation of external sulfate ions in FRA particles, and almost no cracking associated with the precipitation of ettringite or gypsum was observed. This finding supports the hypothesis of the reduction of internal restriction, i.e., the contribution of FRA deformability for a better behavior against expansion due to ESA. The combination of factors affecting internal restriction and ESA kinetics (e.g., specimen shape, type, and degree of exposure, cement matrix) appears to modify the optimum FRA content regarding the resistance in this exposure.

An increase in FRA content leads to an increase in the overall porosity of the mortar, but the net effect depends on whether the predominant influence is on the sulfate penetration rate, the development of internal stresses, or the degree of internal confinement. This ambiguous effect of recycled aggregate content on specimen deterioration has also been observed in other studies in the literature [27,60,61], but further research is necessary for a comprehensive understanding of this phenomenon. It should also be noted that, in this case, only one cement with high  $C_3A$  was evaluated, and the effect of RCA in mixes with other cement types (e.g., different  $C_3A$  content, different  $C_3S/C_2S$  ratio, MCS) could vary.

A few other experiences in the literature showed that mortars or concretes containing recycled aggregates had lower expansion or damage than control mixes, in agreement with the results of the present study. For example, Lee et al [24] studied mortar mixes with 0, 25, 50, 75, and 100% FRA exposed to a sulfate solution. They reported lower expansion of 25 and 50% FRA mixes and higher expansion of 75 and 100% mixes, compared to control mortar. Boudali et al [31] reported a smaller decrease in strength and mass loss in SCC with recycled aggregates compared to control mixes. Santillán et al [27] observed similar or even superior behavior in

719 concretes with coarse recycled aggregates compared to control concretes when subjected to  
720 long-term exposure to sulfate solution. Candamano et al [59] reported a lower loss in  
721 compressive strength of mortars with 100% of FRA in thixotropic mortars. The contribution of  
722 recycled aggregates to reduce internal restriction can provide a plausible hypothesis to explain  
723 all these results.

## 724 **6. CONCLUSIONS**

725  
726 In this paper, multiple instrumental analyses were applied to describe the effects of FRA  
727 content on mortar exposed to different types of sulfate environments. Macro properties such  
728 as expansion, weight change, and visual inspections were performed after 12 months of  
729 exposure. The microstructure of samples was analyzed by thermogravimetric analysis and  
730 microscopic X-ray examination. The following outcomes are stated:  
731

- 732 - Macroscopic and microstructural studies confirmed that the effect of the FRA content  
733 on the behavior of mortar against sulfate attack is not linear with its replacement  
734 percentage. Although a higher permeability to sulfate penetration is attributed to the  
735 higher porosity of the FRA mortars, a positive effect can be attributed to a lower internal  
736 restriction against ESA expansion. The relation between the two opposite effects  
737 seems to have an optimum value that depends on other test parameters. In this case,  
738 the worse behavior corresponds to a replacement of 30% for FRA.
- 739 - In the saturated state, FRA content affects the behavior of cement mortars against  
740 sulfate only when the cement matrix has a poor chemical quality (i.e., a high  $C_3A$   
741 content). In the unsaturated state, cement-based mortars with low  $C_3A$  content showed  
742 a slight effect of the FRA content.
- 743 - The exposure to sulfate-soil showed a higher rate of degradation in samples than  
744 solution exposure. This is due to the combined effect of PSA with pure ESA.
- 745 - The size and shape of specimens had an impact on the kinetics of degradation, in both  
746 solution and soil exposure type.
- 747 - Microscopic analysis showed a lower binding of sulfate ions in the cement paste  
748 attached to FRA particles. This can be related to a good quality in the interface  
749 (disconnected porosity) or to a lower binding capacity of the old cement paste (lower  
750 available portlandite or aluminate phases).
- 751 - The optimal combination of positive and negative effects of FRA is influenced by other  
752 parameters of the test setup, such as the size and shape of the specimens, the type of  
753 exposure, and the properties of the cement matrix. Further research may confirm  
754 whether this phenomenon also affects concrete specimens with coarse recycled  
755 aggregate, causing a different relationship between the internal restriction and ESA  
756 progression.  
757

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762 publication is greatly appreciated.  
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